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Assume that the  
amount of gas is  
constant in the  
following problems. 11.  
A sample of air in a  
syringe exerts a

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pressure of 1.02 atm at 22.0°C. The syringe is placed in a boiling water-bath at 100.0°C. The pressure is

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Molarity is the most  
common concentration  
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Determine the molarity  
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unit in chemistry.

Determine the molarity of a solution where 1.56 moles of NaCl is dissolved in water to give 556 mL of solution. Chapter 13: Solutions - Anoka-Ramsey Community College

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Chapter 13 - Properties of Solutions Solution Composition - a

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Review - most of this section should be a review - solute vs. solvent -- solute is the species that is added to the solution - the more dilute/less concentrated component of a

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are homogeneous  
mixtures of two or  
more pure substances.  
• In a solution, the  
solute is dispersed  
uniformly

### **Chapter 13** **Properties of** **Solutions**

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CHEMICAL

KINETICS418 Solution:

A mechanism that is consistent with the rate law could be:  $X_2 + Y \rightarrow XY + X$  (slow)  $X + Z \rightarrow XZ$  (fast) The rate law only tells us about the slow step. Other mechanisms with different subsequent fast steps are possible. Try to invent one.

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Section 11.1

Assessment page 372

5. Compare the mass of the reactants and the mass of the products in a chemical reaction, and explain how these masses are related. The coefficients in the balanced equation indicate the molar relationship between each

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